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Synthesis and characterization of HAP and beta-TCP ultrafine powders

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AIM OF THE WORK

To obtain ceramic ultra fine powders of HAP and β –TCP, by wet chemical synthesis, from salt solutions.

EXPERIMENTS

- >Hydroxyapatite (HAP) and beta-tricalcium phosphate (β-TCP) powders were obtained by precipitation synthesis, from salt solutions.
- > For experiments was used $(NH_4)_2$ HPO4 and $Ca(NO_3)_2 \cdot 4H_2O$, all made by Fluka.
- > The first step was to obtain the precursor phase of $Ca_9(HPO_4)(PO_4)_5(OH)$, which is named "apatitic tricalcium phosphate", and then converting it to β -TCP by calcinations at low temperature.
- ≻Hydroxyapatite was obtain by thermal treatment at 1000°C for 2hour.
- The formation of the β TCP and the temperature transformation of precursor (apatitic tricalcium phosphate) was evidenced from X-ray diffraction analysis (XRD) and thermal analysis (TG/DTA/DGA). The same characterizations were carried out for analyzing the hydroxyapatite powders.



CONCLUSIONS

• Ceramic nanopowders based on hydroxyapatite and tricalcium phosphate were obtained by precipitation chemical synthesis, from salt solutions (in accordance with ASTM 09-169 files for beta – TCP and ASTM 09-0432 for HAP) • Differential thermal analysis (DTA) indicate an endothermic peak at 790°C, corresponding to the transformation from precursor to β – TCP, in correlation with X-ray diffraction measurements.

• After calcination at 800 °C, when the beta – TCP compound (unique phase) was formed, the average crystallite sizes for nanopowders was of 91.7 nm.

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